

## SUBERIMIDATE CROSSLINKING SHOWS THAT A ROD-SHAPED, LOW CYSTINE, HIGH HELIX PROTEIN PREPARED BY LIMITED PROTEOLYSIS OF REDUCED WOOL HAS FOUR PROTEIN CHAINS

B. AHMADI and P. T. SPEAKMAN

*Department of Textile Industries, The University, Leeds, LS2 9JT, England*

Received 19 July 1978

### 1. Introduction

The cortical cells in  $\alpha$ -keratin fibres like wool are filled with microfibrils, composed of proteins with a low cystine content and a high  $\alpha$ -helix content, embedded in a matrix of high cystine, zero helix protein [1]. The microfibril (diam. 7.5 nm) appears to consist of protein subunits regularly arranged in a superhelix [2]. The structure of the subunit is unknown. It has been suggested that it has two [2] or three [2,3] partially  $\alpha$ -helical protein chains.

A rod-shaped protein ( $2 \times 20$  nm) has been prepared from reduced wool by limited trypsin digestion [4,5]. The protein (merokeratin  $A_1$ ) has a low cystine and a high helix content, and therefore it may be a large fragment of the microfibril subunit. The molecular weight of intact, undenatured  $A_1$  measured in the ultracentrifuge, was compared [5] with the molecular weight of the component chains of  $A_1$  after denaturation, measured by polyacrylamide gel electrophoresis (PAGE). They were unable to say whether  $A_1$  had three or four component chains because of uncertainties in both measurements.

Using the dimethyl suberimide crosslinking technique [6] before denaturation and PAGE, we now show that  $A_1$  consists of four chains. Therefore, if  $A_1$  is a fragment of the microfibril subunit, then the subunit itself may also have four chains.

### 2. Experimental

#### 2.1. Preparation of the merokeratin [4,5]

$A_1$  was prepared at room temperature. Purified

Lincoln wool, 1 g, was reduced in 500 ml 0.2 M thioglycolic acid, 0.1 M  $\text{Na}_2\text{HPO}_4$ , pH 10 for 18 h. The reduced wool was rinsed in 0.1 M  $\text{Na}_2\text{HPO}_4$ , pH 8.0, and digested by 0.02 g trypsin dissolved in 20 ml 0.1 M  $\text{Na}_2\text{HPO}_4$ , pH 8.0. Digestion was inhibited after 1 h by 4 ml 0.25% phenylmethylsulphonylfluoride in 50% v/v isopropanol–water. The digest was clarified at  $50\,000 \times g$  for 30 min, the supernatant was diluted to 750 ml with 0.1 M sodium phosphate, pH 6.0 and the solution was exposed to the air for 22 h to reoxidize cysteine groups to cystine. The reoxidized protein was precipitated at pH 4.8. The precipitate was collected by centrifugation at  $1000 \times g$  for 10 min, redissolved in 5 ml 0.1 M  $\text{Na}_2\text{HPO}_4$ , at pH 8.0, and applied to a Sepharose 6B column.  $A_1$  is the major retarded peak.

#### 2.2. Suberimide crosslinking of the merokeratin [6]

The most concentrated  $A_1$  fraction from the column was diluted to 0.8 mg/ml with 0.1 M  $\text{Na}_2\text{HPO}_4$ , pH 8.0. (Protein concentrations were measured with a Zeiss interferometer.) Eleven 2 ml aliquots of the diluted solution were dialysed against 0.2 M triethanolamine.HCl solutions adjusted to 11 different pH values between 8.2 and 12.4. After dialysis, 2 mg/ml dimethyl suberimide was added to each aliquot, and the pH values were readjusted to the same 11 values. Crosslinking occurred overnight at room temperature. Another 2 ml aliquot of the diluted solution of  $A_1$  was dialysed against triethanolamine.HCl buffer, at pH 8.5, and allowed to stand overnight at room temperature as a control experiment.

### 2.3. Denaturation, PAGE and molecular weight determination [7]

The 11 solutions of crosslinked A<sub>1</sub> and the solution of uncrosslinked A<sub>1</sub> were dialysed against 0.2 M triethanolamine-HCl, pH 8.5, for 2 h. From a solution of 8 M urea, 1% sodium dodecyl sulphate (SDS), 1% 2-mercaptoethanol in distilled water 0.5 ml was added to 0.5 ml of each of the A<sub>1</sub> solutions, and the mixtures were incubated for 6 h at 45°C. After incubation, 20% glycerol and 0.02% bromophenol blue were added to the solutions. From each solution 50 µl was applied to gels prepared using 0.05 M NaH<sub>2</sub>PO<sub>4</sub>, 0.075 M Na<sub>2</sub>HPO<sub>4</sub>, pH 7.2, buffer containing 0.1% SDS, 5% acrylamide, 0.133% *N,N'*-methylene bisacrylamide polymerized by 0.073% ammonium persulphate and 0.11% *N,N,N',N'*-tetramethylethylenediamine. Electrophoresis was carried out using a constant current of 9 mA/gel until the dye front approached the bottom of the tube. The electrophoresis buffer was 0.1 M sodium borate, 0.1 M sodium acetate, 0.1% SDS, pH 8.5. The gels were stained for 30 min at 35°C with 0.25% Coomassie brilliant blue dissolved in 20% trichloroacetic acid, and destained electrolytically in 10% acetic acid in the presence of ion-exchange resin AG 501-X8 (Bio-Rad).

To determine the molecular weight of the uncrosslinked and crosslinked chains from A<sub>1</sub>, alcohol dehydrogenase (yeast) was crosslinked in the same way as A<sub>1</sub>, and a calibration curve was obtained by plotting the mobilities of the uncrosslinked and crosslinked chains from alcohol dehydrogenase and the mobilities of the chains from ovalbumin, pepsin, α-chymotrypsinogen-A, lysozyme, and chymotrypsin, after they had been denatured and electrophoresed in parallel runs with uncrosslinked and crosslinked A<sub>1</sub>, against the logarithm of their molecular weights.

### 3. Results and discussion

When the merokeratin A<sub>1</sub>, not treated with dimethyl suberimidate, is denatured to dissociate it into its component chains, and examined by PAGE on 5% gels there are three bands (fig.1). The molecular weights calculated from the positions of the peaks suggest that the two less intense bands are dimers and trimers of the protein chains in the major band. These small amounts of dimer and trimer may be caused by

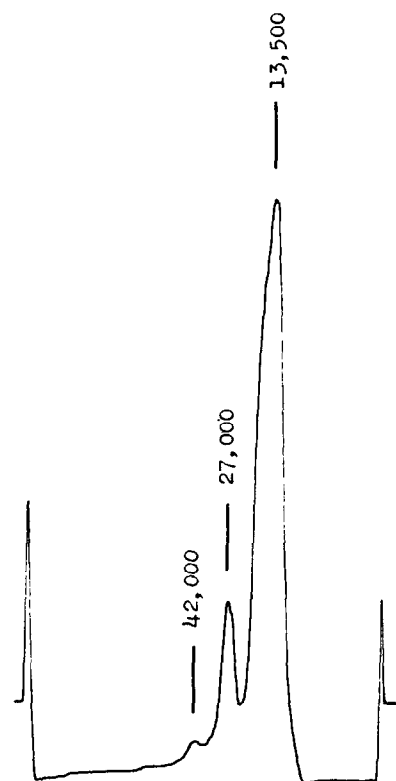


Fig.1. Uncrosslinked merokeratin A<sub>1</sub> denatured and examined by PAGE. Densitometer scan of the gel showing molecular weights calculated from the positions of the peaks.

lanthionine, lysinoalanine or iso-peptide crosslinks between single chains. The crosslinks may be present in the original wool or formed during the preparation of A<sub>1</sub> [8].

The major peak in fig.1 has a shoulder on its heavy side. PAGE of denatured A<sub>1</sub> on 15% gels (not shown) resolves the major peak into two peaks, a larger one (mol. wt 13 500) and a smaller one (mol. wt 15 500). Thus the dimer may consist of three species, mainly mol. wt 27 000 but with smaller amounts of mol. wt 29 000 and 31 000. As expected, the technique did not resolve dimers with these not very different molecular weights. Nor did it resolve the four possible trimers.

Figure 2 is typical of the patterns obtained when A<sub>1</sub> is crosslinked with dimethyl suberimidate at pH values between 9.0 and 11.0, and then denatured and

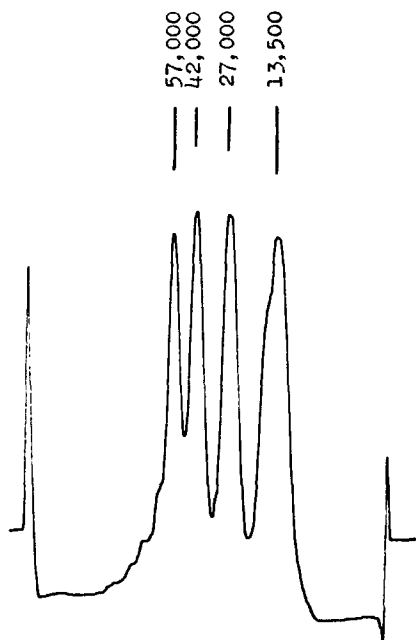


Fig.2. Merokeratin A<sub>1</sub> crosslinked by dimethyl suberimide, at pH 9.9, and denatured and examined by PAGE: molecular weights calculated from the positions of the peaks.

examined by PAGE. Dimethyl suberimide is a bifunctional reagent which can react with two lysine residues. When it reacts with a multichain protein it crosslinks chains in the same multichain aggregate much faster than it crosslinks chains in different aggregates. When four-chain enzymes are crosslinked, denatured and examined by PAGE, patterns like fig.2 are obtained: single chains, dimers and trimers are usually present as well as tetramers [6]. The crosslinking reaction is incomplete probably because there are two reactions competing for the methyl imide groups:

- (i) Reaction with the amine groups of lysine residues.
- (ii) Hydrolysis by water [9].

If one methyl imide group in a dimethyl suberimide molecule reacts with a lysine residue and the other is hydrolysed, then this prevents that lysine residue from taking parts in a crosslink.

When A<sub>1</sub> was crosslinked below pH 9.0, the dimer, trimer and tetramer peaks were smaller than in fig.2 because the lower pH values favour the hydrolysis of the methyl imide groups over their reaction with

lysine residues [9]. When A<sub>1</sub> was crosslinked above pH 11.0, the dimer, trimer and tetramer peaks were again smaller than in fig.2. This suggests that A<sub>1</sub> dissociates into its component chains above pH 11.0.

The mol. wt determined from the positions of the peaks in fig.2 confirm that they are due to A<sub>1</sub> single chains, dimers, trimers and tetramers. The tetramer mol. wt 57 000 is consistent with, but much more precise than, the molecular weight of intact, undenatured A<sub>1</sub> measured in the ultracentrifuge: 50 000–60 000 [4].

These experiments prove that the merokeratin A<sub>1</sub> has four protein chains. The composition and the  $\alpha$ -helix content of A<sub>1</sub> and of the microfibril protein are similar. Therefore the experiments imply that the microfibril subunit may also have four protein chains. In contrast, the subunit of the tonofilaments (also diam. 7.5 nm) in mammalian epidermis has three chains [10], and a merokeratin prepared by limited trypsin digestion of epidermal keratin also has three chains [11].

## References

- [1] Fraser, R. D. B., MacRae, T. P. and Rogers, G. E. (1972) in: *Keratins, their composition, structure and biosynthesis*, pp. 30–120, Thomas, Springfield, IL.
- [2] Fraser, R. D. B., MacRae, T. P. and Suzuki, E. (1976) *J. Mol. Biol.* 108, 435–452.
- [3] Crewther, W. G., Dobb, M. G., Dowling, L. M. and Harrap, B. S. (1968) in: *Symp. Fibrous Proteins, Australia 1967* (Crewther, W. G. ed) pp. 329–340, Butterworth, Sydney.
- [4] Campbell, M. E., Dobb, M. G., Hilburn, M. E., Loh, P., Lotay, S. S., Speakman, P. T., Stainsby, G. and Yarwood, R. E. (1975) in: *Proc. 5th Int. Wool Textile Res. Conf., Aachen, 2*, 243–252.
- [5] Lotay, S. S. and Speakman, P. T. (1977) *Nature* 265, 274–276.
- [6] Davies, G. E. and Stark, G. R. (1970) *Proc. Natl. Acad. Sci. USA* 66, 651–656.
- [7] Weber, K. and Osborn, M. (1975) in: *The Proteins*, 3rd edn (Neurath, H. and Hill, R. L. eds) vol. 1, pp. 179–223, Academic Press, New York.
- [8] Ziegler, K. (1977) in: *Chemistry of Natural Protein Fibers* (Asquith, R. S. ed) pp. 267–300, Wiley, London.
- [9] Peters, K. and Richards, F. M. (1977) *Annu. Rev. Biochem.* 46, 523–551.
- [10] Steinert, P. M., Idler, W. W. and Zimmerman, S. B. (1976) *J. Mol. Biol.* 108, 547–567.
- [11] Skerrow, D., Matoltsy, A. G. and Matoltsy, M. N. (1973) *J. Biol. Chem.* 248, 4820–4826.